

#### According to OSHA Hazard Communication Standard 29 CFR 1910.1200 (GHS)

Product name Azure Alkalinity Booster

Product id AS\_2059\_AZ

Revision date 15/09/2014 Revision: 4

**Supersedes** 10/08/2011

## 1. Identification of the substance & the company

Chemical name Sodium Bicarbonate

Synonym(s) Baking Soda, Bicarbonate of Soda

Molecular weight 84.02

Type of product and use For treatment and balancing of pools, spas and hot tubs

**Supplier** NAVA Water Products

95 MacCorkle Ave. SW, South Charleston, WV 25303,

USA

Toll Free Number: 1-800-811-2327

Emergency Telephone Chemtrec: (800) 424-9300

Medical: (800) 420-9236

#### 2. Hazards identification

GHS Product is not subject to classification according to GHS. No label elements

required.

GHS classification Not classified

Labels and other form of warning Not classified

Symbol(s) Not required

**NFPA Ratings (Scale 0-4)** Health = 0, Fire = 0, Reactivity = 0.

**HMIS Ratings (Scale 0-4)** Health = 0, Fire = 0, Reactivity = 0

# 3. Composition / information on ingredients

Components	CAS No.	Weight %
SODIUM BICARBONATE	144-55-8	100



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#### 4. First-aid measures

**Eye contact** Hold eye open and rinse slowly and gently with water for 15-20 minutes. Remove

contact lenses, if present, after the first 5 minutes, then continue rinsing eye. Get

medical attention immediately.

**Skin contact** Rinse skin immediately with plenty of water for 15-20 minutes. Call a poison

control center or doctor for treatment advice.

**Inhalation** In case of dust inhalation or breathing fumes released from heated material.

remove person to fresh air. Keep him quiet and warm. Apply artificial respiration if

necessary and get medical attention immediately.

**Ingestion** If swallowed, wash mouth thoroughly with plenty of water. Get medical attention

Never give an unconscious person anything to drink.

#### Most important symptoms and effects, acute or delayed

Sodium bicarbonate is a GRAS (Generally Recognized As Safe) food ingredient. No significant toxicity is expected.

- Eye Contact Not irritant

- Skin contact Not irritant

- Inhalation None known

- Ingestion Material is practically non-toxic. Small amounts (1-2 tablespoonfuls) swallowed

during normal handling operations are not likely to cause injury as long as the

stomach is not overly full; swallowing larger amounts may cause injury.

Note to physician Large doses may produce systemic alkalosis and expansion in extracellular fluid

volume with edema. No specific antidote.

Treat symptomatically and supportively.

In case of ingestion DO NOT induce vomiting.

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### 5. Fire - fighting measures

Suitable extinguishing media Material is not combustible. Use extinguishing media appropriate to surrounding

fire conditions.

Unusual fire and explosion

hazards

When heated to decomposition, may release poisonous fumes of Na2O, CO2.

Fire fighting procedure Fire fighters should wear full protective clothing and self-contained breathing

apparatus (SCBA) in positive pressure mode.

#### 6. Accidental release measures

protective clothes

Methods for cleaning up Sweep up and shovel into suitable containers for disposal. Wash spill site with

water after material pickup is complete.

**Environmental precautions** Avoid release to the aquatic environment.

# 7. Handling and storage

Handling Sodium Bicarbonate reacts with acids to yield carbon dioxide gas which can

accumulate in confined spaces. Do not enter confined spaces until they have been well ventilated and carbon dioxide and oxygen levels have been determined to be

safe.

Storage Store in a dry, cool area away from incompatible materials (see "materials to

avoid").



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## 8. Exposure controls / personal protection

#### **Exposure Limits:**

Components	ACGIH-TLV Data	OSHA (PEL) Data
SODIUM BICARBONATE 144-55-8	Not determined	Not determined

**Ventilation requirements** Minimize eye and skin contact by using appropriate protective equipement. Use

local exhaust as necessary, especially under dusty conditions.

Personal protective equipment:

- Respiratory protection Dust mask required if total dust level exceeds 10 mg/m<sup>3</sup>.

- Hand protection Protective gloves

Impervious gloves (rubber or neoprene)

(when working with solutions)

- Eye protection Chemical safety goggles

- **Skin and body protection** Full body protective clothes and boots.

Hygiene measures Do not eat, smoke or drink where material is handled, processed or stored. Wash

hands thoroughly after handling and before eating or smoking. Safety shower and

eye bath should be provided.

## 9. Physical and chemical properties

**Appearance** White crystalline powder

Odor None

**pH** 8.2 (1% solution)

Melting point/range Not applicable (decomposes)

Boiling point/range
Flash point
Evaporation rate (ether=1)
Flammable/Explosion limits
Vapor pressure
Vapor density

Not applicable
Not applicable
Not applicable
Not applicable
Not applicable

- Solubility in water 8.6 g/100ml at 20°C

Auto-ignition temperature Not applicable



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**Bulk density** 62 lb/Ft3 **Specific gravity** 2.20

### 10. Stability and reactivity

Reactivity Reacts with acids

**Stability** Stable

Possibility of hazardous

reactions

Sodium Bicarbonate reacts with acids to yield carbon dioxide gas which can

accumulate in confined spaces.

Conditions to avoid Contact with acids except under controlled conditions. Heating above 65 °C.

Materials to avoid Reacts with acids to release carbon dioxide gas and heat. May yield free caustic in

presence of lime dust (CaO) and moisture (i.e., water, perspiration). Dangerous reaction with monoammonium phosphate or a sodium-potassium alloy may occur.

Hazardous decomposition

products Na2O, CO2

## 11. Toxicological information

#### Acute toxicity:

- **Rat oral LD50** 7.3 g/kg

- Rat inhalation LC50 4.74 mg/l

- Eye irritation (rabbit) Not irritant

- Dermal irritation (rabbit) Not irritant

**Dermal sensitization** Not a sensitizer

Target organ effects None

Chronic toxicity Administration of large doses of sodium bicarbonate to patients with renal

insufficiency can produce systemic alkalosis.

Carcinogenicity Not included in NTP 13th Report on Carcinogens

Not classified by IARC, OSHA, EPA.



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### 12. Ecological information

Aquatic toxicity:

- LC50, Fish 7100 mg/l (Bluegill)

7700 mg/l (Rainbow trout)

- EC50, Crustacea 4100 mg/l (Daphnia)

**Persistence and degradability** Not expected to persist in the environment.

**Biodegradation** Biodegradation is not relevant for inorganic salts.

Bioaccumulative potential Not expected to bioaccumulate

## 13. Disposal considerations

Waste disposal Dispose of in a landfill in accordance with local, state and federal regulations

Disposal of Packaging Empty containers should be disposed of in accordance with all applicable laws

and regulations

## 14. Transportation information

**DOT** Not regulated

# 15. Regulatory information

**USA** Reported in the EPA TSCA Inventory.

- Section 302 (EHS): Not listed

CERCLA/SARA - 302 ext. haz. No

substances

No CERCLA RQ is applicable.



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- SARA 313 Not listed

- SARA (311, 312) Not listed

Canada Listed in DSL

EU Reported in EINECS

**Japan** ENCS no. (1)-164

ISHL no. (1)-164

Australia Listed in AICS

Korea Listed

Philippines Listed in PICCS

#### 16. Other information

# This data sheet contains changes from the previous version in section(s)

2, 4, 5, 7, 8, 10

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In an event of discrepancy between the contents of this SDS and the English version of it, the English version shall prevail.



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End of safety data sheet